A facile approach for the construction of the oxetane ring from 5α -acyloxy- $\Delta^{4(20)}$ -taxoids

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An oxetane ring can be constructed from 5α -acyloxy- $\Lambda^{4(20)}$ -taxoids. The facile intramolecular acyl migration from 5- to 20-position under slightly basic conditions enabled the construction of the oxetane ring in a convenient short cut, whereas the acyl migration from 2- to 20-position left the 2-hydroxyl accessible to a later benzoylation. An unexpected five-membered 4-O, 20-O sulfite ring was formed in the attempted construction of the oxetane ring with 5α -triflate as a leaving group. After the construction of the oxetane ring, treatment with strong base LiHMDS and acetyl chloride gave the expected 4-O-acetate while treatment with acetic anhydride and DMAP gave a 4-O-acetoacetate.

Keywords Taxoids, oxetane ring, intramolecular acyl migration, five-membered 4-0, 20-0-sulfite ring, unusual acylation

Introduction

Paclitaxe (1)¹ has become famous for its unique anticancer mechanism, novel chemical structure and notable antitumor activity. It has been approved for clinical treatment of ovarian cancer and breast cancer respectively by FDA in 1992 and 1994. SAR studies disclosed that an oxetane ring is necessary for its antitumor activity.²

Sinenxan A (2) has been available as a biosynthetic taxane since 1992. Taxinine (3) was separated from the needle of *Taxus X Media*. With these two 5α -acyloxyl- $\Delta^{4(20)}$ -taxoids in hand, one of our efforts to use them as starting material in the search of anticancer agents is to construct the oxetane ring.

Scheme 1

Results and discussion

Sinenxan A could be thoroughly hydrolyzed with methanolic potassium hydroxide to give tetraol 4 in which the 5α -OH is located in the more hindered position than the other hydroxyl groups. Therefore treatment of 4 with acetic anhydride under mild conditions yielded compound 5, along with 2. The $\Delta^{4(20)}$ - double bond of

5 was dihydroxylated with a catalytic amount of OsO_4 and four equivalents of NMO, but the use of Na_2SO_3 during work-up produced a product **7** involving 2- to 20-Ac migration. When less basic reductive agent $NaHSO_3$ was used, the desired compound **6** was obtained in 86% yield. However, the construction of oxetane ring through

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5-triflate 8 failed to afford the desired compound. Two unexpected compounds 9 and 7 were isolated by acidic deprotection of 20-O-TMS with trace amount of cam-

phorsulfonic acid followed by neutralization with saturated $NaHCO_3$ solution.

AcO H
$$\stackrel{\stackrel{\circ}{=}}{\stackrel{\circ}{=}}$$
 H $\stackrel{\circ}{=}$ H

a) KOH, methanol, 60%; b) acetic anhydride, pyridine, CH_2Cl_2 , 25%, 60% (two steps); c) OsO_4 , NMO, 25%, 4 days, 86%; d) TMSCl, pyridine, CH_2Cl_2 , -5%, then TI_2O_3 , -50%, 4 h; e) camphorsulfonic acid, K_2CO_3 , 25%. 9:8%, 7:27% (two steps).

The similarity of the geminal coupling constant (7.8 Hz) of the 20-H to that of paclitaxel $(8.4 \text{ Hz})^{5a}$ aroused our interest. In comparison, 2J of taxoids with a tetrahydrofuran ring at 2-, 3-, 4- and 20- is in the order

of 10 Hz.^{5b} X-ray crystallography of 9 disclosed the unique five-membered sulfite structure where the configuration of the sulfite sulfur was found to be S as shown in Fig. 1.

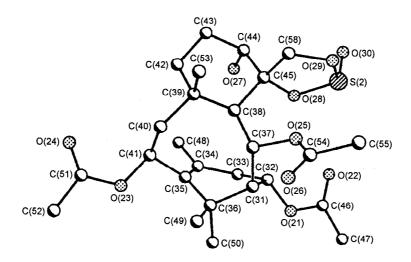


Fig. 1 Structure of compound 9 determined by X-ray crystallography.

Tf₂O was found not to contain SOCl₂ as an impurity, a suspected culprit for sulfite formation. Therefore the formation of the cyclic sulfite 9 from the triflate is rather baffling since it involves fission of the bond be-

tween CF₃- and -SO₂- and reduction of the sulfonate to the sulfite stage.

Another pathway to construct the oxetane ring was from direct osmylation of sinenxan A (Scheme 2).

Treatment of sinenxan A (2) with OsO₄ formed the 1:1 then the 2:1 osmate, ⁶ both tightly bound even under basic conditions, thus excluding the use of catalytic OsO₄ oxidation. Therefore an equimolar amount of OsO₄ was used without NMO for its oxidative regeneration. The 1:1 osmate thus formed was cleaved reductively with Na₂SO₃. This slightly basic medium again gave rise to the acyl migration, from 2- and 5-position to 20-OH, furnishing 12 and 13, alongside of 11 as the unrear-

ranged product. The formation of 12, as the result of 5-to 20-migration, has to involve the intermediacy of 4-OAc since the 5-axial ester and 4-axial-CH₂OH are too far apart. This spatial disposition can be readily seen from the positive NOE between 20-proS and the angular Me protons. The newly discovered 5- to 20-acyl migration obviates the inconvenient protection and deprotection of 20-OH with a TMS group.⁷

Scheme 2

a) OsO₄, 25—30°C, 1 h, Na₂SO₃; b) MsCl, pyridine, 25°C, 60 h, 74%; c) K₂CO₃, methanol, 0°C, 15 min. d) DBU, toluene, 110°C, 1 h, 87% (two steps).

Conversion of the 5α -OH of compound 12 to its methanesulfonate 14 was uneventful since it is the only unhindered OH. This was followed by removal of the protective acyl group at the 20-position. It was fortunate indeed (*vide infra*) to have a simultaneous hydrolysis of the 2-ester (giving 15) as the result of the facile 2- to 20-migration. Oxetane ring closure was accomplished in the usual way to give 16 by heating 15 with DBU.

When the above method for construction of the oxetane ring was applied to taxinine (3) (Scheme 3), a similar result was obtained but in the osmylation-hydrolysis-migration procedures the ratio of 5-migrated product to 2-migrated product (1:4) was much lower than that from sinenxan A (1:1) (Scheme 2). This is reasonable since the cinnamoyl carbonyl is comparatively more reluctant to receive a nucleophile. To overcome the low yield caused by unsatisfactory migration, the compounds 17 and 18 were subjected to hydrolysis, then followed by careful reacetylation to afford 21 in 52% yield which served as an improved version of 19. The target compound 25 was obtained from 19 or 21 using the similar

strategy described in Scheme 3.

Because the 4-hydroxyl group of 4-deacetyl paclitaxel is located in a very hindered position, its reacetylation has to be effected under vigorous conditions and usually gives only moderate yields. The attempted 4-acetylation of the compound 26¹⁰ derived from 2, under similar forcing conditions, led to the formation of a 4-O-acetoacetate (27) as a major product (Scheme 4). With strong base to form its 4-oxide anion, the desired product 29 was obtained only in low yield, still accompanied with the formation of 27.

The formation of 4-O-acetoacetate is unusual. Evidently the 4-OH is too hindered to allow fruitful encounters with the bulky Ac_2O to form the 4-O-acetate. There must be a more reactive species which gives directly the acetoacetate. We would like to propose 28 as the less bulky reactive species, which is generated during the prolonged acetylation process. The involvement of 28 is strongly supported by the presence of the enolic form as the major isomer (ca. 60%) in the acetoacetate (27).

Scheme 3

a) OsO_4 , THF, 25%, 14 h, Na_2SO_3 , 17 46%, 18 26%, 19 12%; b) K_2CO_3 , methanol, 0%, 83%; c) acetic anhydride, pyridine, 25%, 48 h, 52%; d) MsCl, pyridine, 25%, overnight, 83%; e) MsCl, pyridine, 25%, 24 h, 72%; f) K_2CO_3 , methanol, 0%, 85%; g) DBU, toluene, 105%, 1.5 h, 46%.

Scheme 4

a) Ac_2O , DMAP, 35 °C, 5 h, 56%; b) LiHMDS, AcCl, 0°C, 1.5 h, 51%.

When 27 was dissolved in slightly acidic CDCl₃, a slow conversion of the enolic form to the keto form could be observed by ¹H NMR. This is indicative of the kinetically controlled acylation of 26 by some enolic acylating agent, here proposed as 28. Diketene, an isomer of 28, can be safely excluded as a likely acylating agent by having the *wrongly* disposed terminal double bond.

There is no satisfactory rationale for step-wise and site-specific *bis*-acetylation where one-step acylation by agent like **28** is not invoked. Further support for one-step acylation was furnished by the following experiment. A mixture of Ac₂O and DMAP was allowed to stand for one to two days, to which more than one equivalent of aniline was added all at once. The expected PhNHCO-CH₂-COCH₃ formed in trace amounts was not amenable to convenient separation or direct detection in

the presence of acetanilide, the major product. It was derivatized by H_2N -O- CH_2 -COOH to give an oxime, which can be extracted into aqueous alkali, isolated and compared with an authentic specimen of PhNH-CO- $CH_2C(CH_3) = N$ -O- CH_2 -COOH (TLC and 1H NMR).

Conclusions

A facile approach was established for the construction of the oxetane ring using an intramolecularly available acyl group for the protection of 20-OH by an unexpected 5- to 20-migration. This "internal protection" obviates the inconvenient protection and deprotection of 20-OH. It is promising to be a general procedure for the construction of the oxetane ring from 5α -acyloxy- $\Delta^{4(20)}$ -taxoids. The simultaneous removal of the 2-acyl group

accompanied by hydrolysis of 20-ester facilitated 2-modification, for example, 2-benzoylation as a mimicry of paclitaxel, where the 2-benzoate is another essential contributor to activity. This avoids the undesirable hydrolysis step which can hardly be accomplished without endangering other ester groups.

The unusual one step acylation of 4-OH to give 4-O-acetoacetate was observed. The possible mechanism was confirmed by 1H NMR and chemical studies.

Experimental

General procedures

All reactions were carried out under nitrogen atmosphere. Anhydrous THF was freshly distilled from LiAlH₄. All solutions used in the workup procedures are saturated unless otherwise specified. Silica gel (180—200 mesh) was used for column chromatography. NMR spectra were recorded on Bruker AM500 and ARX400 instruments and tetramethylsilane was used as internal reference. Mass spetra were obtained on a Jeol JMS-SX102 mass spectrometer (matrix: *m*-NBA *m*-nitro benzyl alcohol). High resolution mass spectra were recorded on a Bruker FTMS-APEXTM II mass spectrometer (matrix: *m*-NBA). Melting points are uncorrected.

The structure of sinenxan A was established by X-ray crystallography and the assignment of ¹H signals was firmly ascertained by 2D NMR. The characteristic splitting patterns of individual groups were served as a convenient point of departure for following changes of functionalities.

Preparation of compound 5 A solution of 2 (22.95 g, 45.50 mmol) in methanol (200 mL) was treated with KOH (12.76 g, 0.228 mol, in 18 mL of H₂O) at 60°C for 7 h. After solvent was removed, the residue was dissolved in pyridine (20 mL) and acetic anhydride (46.5 mL, 0.492 mol) divided in four portions was added in a reaction period of 8 days at 25%, then the excess acetic anhydride was decomposed by addition of methanol (50 mL). After removal of methanol, the mixture was poured into ice-water (400 mL), extracted with ethyl acetate (4 × 150 mL). The combined extracts were washed with dilute HCl (10% HCl in 1000 mL of H₂O), brine, dried over Na₂SO₄, concentrated and the residue was chromatographed (petroleum ether/ethyl acetate 3.5:1) to give recovered 2 (16.45

g, 72%) and 5 (3.57 g, 60% based on recovery of starting material). 5: mp 71—73°C. ¹H NMR (500 MHz, CDCl₃): δ 6.11(dd, J = 12.0, 5.6 Hz, 1H, 10-H), 5.34(dd, J = 6.3, 2.3 Hz, 1H, 2-H), 5.10 (s, 1H, 20-H), 5.03 (dd, J = 9.3, 4.6 Hz, 1H, 14-H), 4.78(t, J = 1.5 Hz, 1H, 20-H), 4.18(t, J)= 2.7 Hz, 1H, 5-H), 2.77 (dd, J = 19.0, 9.3 Hz,1H, 13-H), 2.37—2.29(m, 2H, 13-H and 9-H), 2.21 (d, J = 6.2 Hz, 1H, 3-H), 2.09 (m, 1H, 6-H), 2.10(s, 3H, Ac), $2.08(s, 3H, 18-CH_3)$, 2.04(s, 3H, Ac), 2.02(s, 3H, Ac), 1.84(d, J = 2.1)Hz, 1H, 1-H), 1.69-1.80(m, 2H, 6-H and 7-H), $1.65(s, 3H, 16-CH_3), 1.59(dd, J = 14.8, 5.5 Hz,$ 1H, 9-H), 1.08-1.19(m, 1H, 7-H), 1.11(s, 3H,17-CH₃), 0.81(s, 3H, 19-CH₃). Anal. $C_{26}H_{38}O_7$. Calcd: C, 67.51; H, 8.28. Found: C, 67.80; H, 8.41.

Preparation of compound 6 A solution of 5 (700 mg, 1.513 mmol), NMO monohydrate (850 mg, 6.289 mmol) in a mixed solvent (THF 200 mL, acetone 30 mL, H₂O 6 mL) was treated with OsO₄ aqueous solution (4%, 0.9 mL, 0.146 mmol) at 25℃ for 4 days, then aqueous NaHSO3 was added. After being concentrated, the residue was taken up by ethyl acetate and the organic phase was washed successively with 1 N HCl, aqueous NaHCO₃ solution, aqueous NaHSO₃ solution, brine, dried (Na₂SO₄) and concentrated to give crude product which was purified by column chromatography to yield product 6 as a white powder (643 mg, 86%). mp 179—182°C. ¹H NMR (500 MHz, CD Cl_3): $\delta 6.03$ (dd, J = 12.1, 5.5 Hz, 1H, 10-H), 5.64(dd, J = 9.3, 4.7 Hz, 1H, 14-H), 5.33(dd, J)= 5.4, 2.2 Hz, 1H, 2-H), 3.75(t, J = 2.7 Hz,1H, 5-H), 3.51(d, J = 10.8 Hz, 1H, 20-H), 3.48(d, J = 10.8 Hz, 1H, 20-H), 2.68(d, J = 5.4 Hz,1H, 3-H), 2.63 (dd, J = 18.9, 9.3 Hz, 1H, 13-H), 2.52(dd, J = 18.9, 4.7 Hz, 1H, 13-H), 2.25(dd, J = 14.9, 12.3 Hz, 1H, 9-H), 2.13(s, 3H,Ac), 2.05(s, 3H, 18-CH₃), 2.04(s, 3H, Ac), 2.03(s,3H,Ac),2.04-1.96(m,1H,6-H),1.80-1.72 (m, 2H, 6-H and 7-H), 1.76 (d, J = 2.2Hz, 1H, 1-H), $1.63(s, 3H, 16-CH_3)$, 1.42(dd, J)= 14.9, 5.5 Hz, 1H, 9-H), 1.13(s, 3H, $17-CH_3$), $1.01-1.07(m, 1H, 7-H), 0.76(s, 3H, 19-CH_3).$ FAB MS m/z: 497(M + H⁺). HR ESI MS Calcd. for $C_{26}H_{40}O_9(M+H^+)$: 497.2751. Found: 497.2775.

Preparation of compound 8 To a solution of 6

(105 mg, 0.211 mmol) and pyridine (300 mg, 3.793 mmol) in CH₂Cl₂(3 mL) was added trimethylsilyl chloride (TMSCl, 118 mg, 1.086 mmol) at -5% and the reaction mixture was stirred at -5% for 20 min, then the solvent was removed in vacuo and the residue was redissolved in CH₂Cl₂. This solution was cooled to -50 to -60°C and i-Pr₂NEt (480 mg, 3.714 mmol) and triflic anhydride (Tf₂O, 531 mg, 1.882 mmol) were added. The resulting mixture was stirred for 4 h when the temperature was allowed to warm-up to room temperature, then poured into a mixture of ice-water/cther/ NaHCO₃ solution. The organic layer was separated and the water layer was extracted with ether. The combined organic layers were washed with dilute HCl (3 N, 3 mL in 50 mL of H₂O), aqueous NaHCO₃, brine and dried over Na₂SO₄. After concentration, the crude was passed through a silica gel column, which was eluted quickly with petroleum ether/ethyl acetate (3:1) to give impure product (210 mg) which was taken to the next step without further purification. ¹H NMR (500 MHz, CD- Cl_3): $\delta 6.03$ (dd, J = 12.2, 5.3Hz, 1H, 10-H), 5.72 (dd, J = 9.4, 3.8 Hz, 1H, 14-H), 5.27(dd, J =4.9, 2.4 Hz, 1H, 2-H), 4.19(s, 1H, 5-H), 3.65 (d, J = 10.3 Hz, 1H, 20-H), 3.37(d, J = 10.3)Hz, 1H, 20-H), 3.10(dd, J = 19.6, 9.4 Hz, 1H,13-H), 2.60(d, J = 4.9 Hz, 1H, 3-H), 2.40(br.d, J = 19.6 Hz, 1H, 13-H), 2.27(dd, J = 15.2, 12.2 Hz, 1H, 9-H), 2.09(s, 3H, Ac), 2.04(s, 3H, Ac), 2.03(s, 3H, 18-CH₃), 2.01(s, 3H, Ac),2.00-1.82 (m, 3H, 2×6 -H and 7-H), 1.65 (d, J $= 2.4 \text{ Hz}, 1H, 1-H), 1.60(s, 3H, 16-CH_3), 1.41$ (dd, J = 15.2, 5.4 Hz, 1H, 9-H), 1.15(d, J =11.6 Hz, 1H, 7-H), 1.11(s, 3H, 17-CH₃), 0.79 (s, 3H, 19-CH₃).

Compounds 7 and 9 A solution of 8 (210 mg) from the above step in methanol (5 mL) was treated with a trace amount of camphorsulfonic acid at 25 °C until the starting material disappeared, then one drop of NaHCO₃ solution was added and the solvent was removed in vacuo. The residue was taken up with ether, washed with brine, dried over Na₂SO₄, concentrated and the residue was chromatographed (petroleum ether/ethyl acetate 1.8:1 to 1:1) to yield 7 (31 mg, 27 %), and 9 (10 mg, 8%). 7: mp 91—94°C; ¹H NMR (500 MHz, CDCl₃): δ 6.01 (dd, J = 12.2, 5.5 Hz, 1H, 10-H), 5.63(t, J = 7.1 Hz, 1H, 14-H), 4.81(d, J = 11.5 Hz, 1H, 20-H), 4.38(d, J = 11.5 Hz, 1H,

20-H), 4.07(dd, J = 5.2, 2.3 Hz, 1H, 2-H), 2.59(d, J = 7.1 Hz, 2H, 13-H), 2.41(d, J = 5.2 Hz, 1H, 3-H), 2.18 (dd, J = 14.7, 12.3 Hz, 1H, 9-H), 2.08(s, 3H, Ac), 2.07(s, 3H, Ac), 2.04(s, 3H, Ac) $3H, 18-CH_3$, 2.02(s, 3H, Ac), 1.97(dd, J = 10.5,5.8 Hz, 1H,6-H), 1.89(d, J = 2.3 Hz, 1H, 1-H), 1.71—1.81(m, 2H, 6-H and 7-H), 1.54(s, 3H, $16-CH_3$), 1.37(dd, J = 14.7, 5.4 Hz, 1H, 9-H), $1.13(s, 3H, 17-CH_3), 1.03(br.d, J = 13.0 Hz,$ 1H, 7-H), 0.91(s, 3H, 19-CH₃). HR ESI MS Calcd . for $C_{26} H_{40} O_9$ (M + Na⁺) : 519 . 2570 . Found : 519.2601. **9**: ¹H NMR (500 MHz, CDCl₃): δ 6.06 (dd, J = 12.1, 5.4 Hz, 1H, 10-H), 5.63(dd, J =9.5, 4.1 Hz, 1H, 14-H), 5.30(dd, J = 4.9, 2.2 Hz, 1H, 2-H), 4.31(d, J = 7.8 Hz, 1H, 20-H), 4.24(d, J = 7.8 Hz, 1H, 20-H), 4.08(br. s, 1H,5-H), 3.02(d, J = 4.9 Hz, 1H, 3-H), 2.68(dd, J)= 19.2, 9.5 Hz, 1H, 13-H), 2.48(br.d, J = 19.2Hz, 1H, 13-H), 2.29(dd, J = 14.0, 11.3 Hz, 1H,9-H), 2.10-2.05 (m, 1H, 6-H), 2.06 (s, 3H, Ac), $2.05(s, 3H, 18-CH_3)$, 2.04(s, 3H, Ac), 2.02(s, 3H, Ac), 1.88-1.94(m, 1H, 6-H), 1.76(s,1H, 1-H), 1.66-1.74(m, 1H, 7-H), 1.63(s, 3H, 1.63) $16-CH_3$, 1.45 (dd, J = 14.9, 5.3 Hz, 1H, 9-H), $1.12(s, 3H, 17-CH_3), 1.11-1.07(m, 1H, 7-H),$ $0.77(s, 1H, 19-CH_3)$.

Preparation of compounds 11, 12 and 13 To a solution of 2 (800 mg, 1.585 mmol) in THF (40 mL) was added OsO₄ water solution (4%, 10.24 mL, 1.649 mmol, diluted in 120 mL of THF) in a period of 1 h. The solution was stirred at 25-30°C for 4 h. Na₂SO₃ (1.6 g, 12.7 mmol) was dissolved in water and added to the stirred reaction mixture. After 6 h, another portion of Na₂SO₃(0.8 g, 6.35 mmol) was added and the mixture was stirred for 4 more hours, then neutralized with dilute HCl (10%) and concentrated. The ethyl acetate extracts $(5 \times 20 \text{ mL})$ of the residue were washed with water, dried over Na2SO4, concentrated and the residue was chromatographed (petroleum ether/acetone 9:1 to 5:1) to give 11 (0.13 g, 15 %), 12 (0.28 g, 32 %), **13** (0.26 g, 31 %). **11**: mp 168—170℃; ¹H NMR (400 MHz, CDCl₃): δ 6.00(dd, J = 12.1, 5.5 Hz, 1H, 10-H), 5.61 (dd, J = 9.0, 4.6 Hz, 1H, 14-H), 5.32(dd, J = 5.2, 2.2 Hz, 1H, 2-H), $5.19(br.s, 1H, 5-H), 3.52(s, 2H, 2 \times 20-H),$ 2.67(dd, J = 19.1, 9.0 Hz, 1H, 13-H), 2.57(dd,J = 19.1, 4.6 Hz, 1H, 13-H), 2.55(d, J = 5.1

Hz, 1H, 3-H), 2.29(dd, J = 14.9, 12.3 Hz, 1H,9-H), 2.17(s, 3H, Ac), 2.13(s, 3H, Ac), 2.10 $(s, 3H, 18-CH_3), 2.04(s, 3H, Ac), 2.01(s, 3H,$ Ac), 1.90-1.82 (m, 2H, 6-H), 1.79 (d, J = 2.1Hz, 1H, 1-H), 1.68-1.76 (m, 1H, 7-H), 1.62 $(s, 3H, 16-CH_3), 1.44(dd, J = 14.9, 5.4 Hz, 1H,$ 7-H), 0.79(s, 3H, 19-CH₃). HR ESI MS Calcd. for $C_{28}H_{42}O_{10}(M + Na^+)$: 561.2676. Found: 561.2695. 12: mp67-70 °C; ¹ H NMR (400 MHz, CDCl₃): δ 6.04(dd, J = 12.2, 5.5 Hz, 1H, 10-H), 5.68(dd,J = 9.4, 4.6Hz, 1H, 14-H), 5.33 (dd, J = 5.2, 2.2Hz, 1H, 2-H), 4.44(d, J = 11.9 Hz, 1H, 20-H), 4.08(d, J = 11.9 Hz, 1H, 20-H), 3.86(t, J = 2.7)Hz, 1H, 5-H), 2.70(d, J = 5.2Hz, 1H, 3-H), 2.62(dd, J = 19.1, 8.2 Hz, 1H, 13-H), 2.49(br.d, J = 19.1 Hz, 1H, 13-H), 2.26 (dd, J = 14.9, 12.2 Hz, 1H, 9-H), 2.19(s, 3H, Ac), 2.09(s, $3H, Ac), 2.05(s, 3H, 18-CH_3), 2.05(s, 3H, Ac),$ 2.02(s, 3H, Ac), 2.02-1.97(m, 1H, 6-H), 1.78(d, J = 2.2 Hz, 1H, 1-H), 1.85—1.73(m, 2H, 6-H and 7-H), 1.63(s, 3H, 16-CH₃), 1.44(dd, J =14.9, 5.4 Hz, 1H, 9-H), 1.12(s, 3H, $17-CH_3)$, $1.06(m, 1H, 7-H), 0.83(s, 3H, 19-CH_3)$. FAB MS m/z: 537 (M + H⁺); HR ESI MS Calcd. for $C_{28}H_{42}$ - $O_{10}(M + Na^{+})$: 561.2676. Found: 561.2675. **13**: mp 154—156 °C; ¹ H NMR (400 MHz, CDCl₃): δ 5.99(dd, J = 12.2, 5.4 Hz, 1H, 10-H), 5.72(dd,J = 7.6, 6.2 Hz, 1H, 14-H), 5.06(d, J = 3.1 Hz, 1H, 5-H), 4.85(d, J = 11.3 Hz, 1H, 20-H), 4.46(d, J = 11.3 Hz, 1H, 20-H), 4.12-4.01 (m, 1H, 2-H), 4.00(s, 1H, OH), 3.06(br.s, 1H, OH), 2.58(br. d, J = 7.7 Hz, 2H, 2×13 -H), 2.24(dd, J = 14.8, 12.2 Hz, 1H, 9-H), 2.21(d, J = 4.2)Hz, 1H, 3-H), 2.17(s, 3H, Ac), 2.13(s, 3H, Ac), 2.10(s, 3H, Ac), $2.06(s, 3H, 18-CH_3)$, 2.05 $(s, 3H, Ac), 1.92-1.82(m, 3H, 2 \times 6-H \text{ and } 1-$ H), 1.77-1.66 (m, 1H, 7-H), 1.54 (s, 3H, 16- CH_3), 1.41(dd, J = 14.8, 5.4Hz, 1H, 9-H), 1.22-1.14(m, 1 H, 7-H), 1.13(s, 3H, 17-CH₃), 0.96(s, 3H, 19-CH₃). HR ESI MS Calcd. for C₂₈H₄₂O₁₀ $(M + Na^{+})$: 561.2676. Found: 561.2689.

Preparation of compound 14 A solution of 12 (260 mg, 0.483 mmol) in pyridine (6 mL) was treated with methanesulfonyl chloride (150 mL, 1.941 mmol) at room temperature for 60 h, then poured into ice-water

(50 mL), and the mixture was extracted with ethyl acetate $(3 \times 50 \text{ mL})$. The combined organic layers were washed with dilute HCl (1N, 60 mL), brine and dried over Na2SO4, concentrated and the residue was chromatographed to yield product (220 mg, 74%) as a white solid, mp 199—202 $^{\circ}$ C. ¹H NMR (400 MHz, $CDCl_3$): $\delta 6.03(dd, J = 12.1, 5.5 Hz, 1H, 10-H),$ 5.67(dd, J = 9.3, 4.4 Hz, 1H, 14-H), 5.34(dd, J)= 4.9, 2.1 Hz, 1H, 2-H), 4.85(s, 1H, 5-H), 4.65(d, J = 12.2 Hz, 1H, 20-H), 3.99 (d, J = 12.2)Hz, 1H, 20-H), $3.09(s, 3H, -SO_2CH_3)$, 2.66(dd, 3H)J = 18.7, 9.3 Hz, 1H, 13-H), 2.62(d, J = 4.9Hz, 1H, 3-H), 2.50(dd, J = 18.7, 4.4 Hz, 1H,13-H), 2.27 (dd, J = 14.8, 12.2Hz, 1H, 9-H), 2.23(s, 3H, Ac), 2.10(s, 3H, Ac), 2.08(s, 3H, $18-CH_3$, 2.05(s, 3H, Ac), 2.04(s, 3H, Ac), 2.03 -1.91 (m, 3H, 2×6 -H and 7-H), 1.79 (d, J = 2.1Hz, 1H, 1-H), $1.63(s, 3H, 16-CH_3)$, 1.49(dd, J)= 14.8, 5.5 Hz, 1H, 9-H), 1.28-1.22(m, 1H, 7-H), $1.12(s, 3H, 17-CH_3)$, $0.86(s, 3H, 19-CH_3)$. FAB MS m/z: 617(M + H⁺); HR ESI MS Calcd. for $C_{29} H_{44} O_{12} S(M + Na^{+}) 639.2451$. Found: 639.2481.

Preparation of compound 16 A solution of 14 (300 mg, 0.486 mmol) in methanol (30 mL) was treated with K_2CO_3 (134 mg in 2 mL of H_2O_3 , 0.970 mmol) at 0℃ for 15 min, then dilute HCl (1 N, 1 mL) was added, and the mixture was concentrated. The residue was taken up with ethyl acetate, washed with brine, dried over Na₂SO₄ and concentrated to give crude 15, which was taken to the next step without purification. ¹H NMR (400 MHz, CDCl₃): δ 5.99 (dd, J =12.1, 5.4 Hz, 1H, 10-H), 5.65(dd, J = 8.9, 4.9)Hz, 1H, 14-H), 4.72(d, J = 2.5 Hz, 1H, 5-H), 4.24(d, J = 11.4Hz, 1H, 20-H), 4.08(dd, J =5.2, 2.2 Hz, 1H, 2-H), 3.53(d, J = 11.4 Hz, 1H, 20-H), $3.08(s, 3H, -SO_2CH_3)$, 2.64-2.60(m, 2H, 13-H), 2.37(d, J = 5.2 Hz, 1H, 3-H),2.17(m, 1H, 9-H), 2.16(s, 3H, 18-CH₃), 2.06 (s, 3H, Ac), 2.02(s, 3H, Ac), 2.00-1.88(m,3H, 2×6 -H and 7-H), 1.84(d, J = 2.2 Hz, 1H, 1-H), $1.51(s, 3H, 16-CH_3)$, 1.41(dd, J = 14.9, 5.4)Hz, 1H, 9-H), 1.21-1.16(m, 1H, 7-H), 1.12(s, $3 \text{ H}, 17\text{-CH}_3$, $0.88(s, 3H, 19\text{-CH}_3)$. FABMS m/z: $533 (M + H^{+}).$

To a solution of **15** in toluene (20 mL) was added DBU (150 mg, 0.970 mmol) and the reaction mixture

was stirred at 110℃ for 1 h, then cooled to room temperature, applied directly to column chromatography (petroleum ether/ethyl acetate 3:7) to yield product 16 as a white solid (185 mg, 87% from 14), mp 162— 164°C. ¹H NMR (400 MHz, acetone- d_6): δ 5.95(dd, J = 12.4, 5.5 Hz, 1H, 10-H), 5.60 (dd, J = 9.1, 4.8 Hz, 1H, 14-H), 4.64 (dd, J = 8.9, 2.6 Hz, 1H, 5-H), 4.59(d, J = 8.0 Hz, 1H, 20-H), 4.30(d, J = 8.0 Hz, 1H, 20-H), 4.08(dd, J = 5.5, 2.8)Hz, 1H, 2-H), 2.85(d, J = 5.5Hz, 1H, 3-H), 2.57(dd, J = 18.9, 9.1 Hz, 1H, 13-H), 2.47(dd,J = 18.9, 4.8 Hz, 1H, 13-H), 2.33(dd, J = 14.7, 12.4 Hz, 1H, 9-H), 2.09(s, 3H, Ac), 2.00(s, 18- CH_3), 1.88(s, 3H, Ac), 1.88—1.85(m, 2H, 6-H), 1.74(d, J = 5.4 Hz, 1H, 1-H), 1.71-1.60(m, 1H, 7-H), 1.58(s, 1H, 19-CH₃), 1.45-1.51(m, 2H, 9-H and 7-H), 1.32(s, 3H, 17-CH₃),1.13(s, 3H, 16-CH₃). ¹³C NMR (125 MHz, CD- Cl_3): δ 172.4, 170.0, 135.9, 134.4, 87.7, 82.2, 76.8, 70.6, 70.3, 70.2, 61.2, 47.8, 45.2, 37.8, 37.6, 36.3, 35.8, 31.9, 27.4, 25.8, 21.9, 21.6, 21.3, 20.9: HR FAB MS Calcd. for $C_{24}H_{36}O_7(M +$ H⁺): 437.2534, Found: 437.2539.

Prepatation of compounds 17, 18 and 19 A solution of 3 (66 mg, 0.109 mmol) in THF (5 mL) was treated with aqueous $OsO_4(4\%, 0.70 \text{ mL}, 0.109 \text{ mmol})$ at 25% for 14 h, then aqueous $Na_2SO_3(5\%, 1 \text{ mL}, 0.397 \text{ mmol})$ was added and stirred at 25% for 2 h. The mixture was neutralized with HCl (5%) and concentrated in vacuo. The residue was extracted with ethyl acetate (4 × 10 mL). The combined organic layers were washed with brine, dried over Na_2SO_4 , concentrated and the residue was purified by preparative TLC (silica gel, CH_2Cl_2/CH_3OH 35:1) to give 17 (32 mg, 46%), 18 (18 mg, 26%), and 19 (8 mg, 12%).

17: ¹H NMR (400 MHz, CDCl₃): δ 7.75—7.64(m, 3 H, COCH = CHPh and Ph - H), 7.47—7.38(m, 3H, Ph-H), 6.47(d, J = 15.9 Hz, 1H, COCH = CHPh), 6.00(d, J = 10.3 Hz, 1H, 10-H), 5.75(d, J = 10.3 Hz, 1H, 9-H), 5.13(t, J = 3.0 Hz, 1H, 5-H), 4.80(d, J = 11.4 Hz, 1H, 20-H), 4.55(d, J = 11.4 Hz, 1H, 20-H), 4.23(br. s, 1H, 2-H), 4.07(br. d, J = 11.0 Hz, 1H, 2-OH), 3.28(s, 1H, 4-OH), 3.03(d, J = 19.9 Hz, 1H, 14-H), 2.73(dd, J = 19.9, 6.7 Hz, 1H, 14-H), 2.69(d, J = 4.4 Hz, 1H, 3-H), 2.34(s, 3H, Ac), 2.30

(dd, J = 6.7, 1.7 Hz, 1H, 1-H), 2.15(s, 3H, Ac), 2.09(s, 3H, Ac), 2.05(s, 3H, 18-CH₃), 1.92—1.86(m, 2H), 1.76(dt, J = 13.6, 3.0 Hz, 1H, 6-H), 1.68(s, 3H, 16-CH₃), 1.68—1.59(m, 1H), 1.17(s, 3H, 17-CH₃), 0.99(s, 3H, 19-CH₃).

18: ¹H NMR (400 MHz, CDCl₃): δ 7.73— 7.62 (m, 3H, COCH = CHPh and Ph-H), 7.44— 7.35 (m, 3H, Ph-H), 6.48 (d, J = 15.9 Hz, 1H, COCH = CHPh), 6.02(d, J = 10.3 Hz, 1H, 10-H), 5.78(d, J = 10.3 Hz, 1H, 9-H), 5.56(dd, J)= 5.0, 2.1 Hz, 1H, 2-H), 5.15(t, J = 3.0 Hz,1H, 5-H), 4.12(d, J = 11.4 Hz, 1H, 20-H), 3.78(d, J = 11.4 Hz, 1H, 20-H), 3.03 (d, J = 19.8Hz, 1H, 14-H), 2.73(dd, J = 19.8, 6.7 Hz, 1H,14-H), 2.65(d, J = 5.0 Hz, 1H, 3-H), 2.30(s,3H, Ac), 2.25(dd, J = 6.7, 1.7Hz, 1H, 1-H),2.12(s, 3H, Ac), 2.08(s, 3H, Ac), 2.06(s, 3H, $18-CH_3$), 1.91-1.86(m, 2H), 1.78-1.75(m, 2H)1H, 6-H), $1.68(s, 3H, 16-CH_3), 1.18-1.58(m,$ 1H), $1.16(s, 3H, 17-CH_3), 0.90(s, 3H, 19-CH_3).$

¹H NMR (400 MHz, CDCl₃): δ 7.70 (d, J = 16.0 Hz, 1H, COCH = CHPh), 7.59—7.48 (m, 2H, Ph-H), 7.44-7.34(m, 3H, Ph-H), 6.43(d, J = 16.0 Hz, 1H, COCH = CHPh), 6.04(d, J= 10.3 Hz, 1H, 10-H), 5.79(d, J = 10.3 Hz, 1H,9-H), 5.63(dd, J = 4.9, 1.9 Hz, 1H, 2-H), 4.66(d, J = 12.0 Hz, 1H, 20-H), 4.18(d, J = 12.0Hz, 1H, 20-H), 3.89(t, J = 2.6Hz, 1H, 5-H), 3.30(br. s, 1H, 5-OH), 3.17(d, J = 19.7 Hz, 1H,14-H), 3.12(d, J = 4.9 Hz, 1H, 3-H), 2.73(dd,J = 19.7, 6.8 Hz, 1H, 14-H), 2.23(s, 3H, Ac),2.20(s, 3H, Ac), 2.15(dd, J = 6.8, 1.9 Hz, 1H,1-H), 2.09(s, 3H, Ac), 2.03(s, 3H, 18-CH₃), 1.91-1.82(m,1H), 1.77(s,3H,16-CH₃), 1.74-1.57 (m, 3H), 1.15 (s, 3H, 17-CH₃), 0.89 (s, 3H, 19-CH₃).

Preparation of compound 20 A solution of a mixture of 17 and 18 (260 mg , 0.4058 mmol) in methanol (26 mL) was treated with 1N aqueous K_2CO_3 (4.06 mL, 2.029 mmol) at 0°C for 4.5 h. The reaction was quenched with 5% aqueous HCl (pH 7). The resulting solution was concentrated in vacuo and the residue was dissolved in ethyl acetate (60 mL). The organic layer was washed with brine , dried (Na₂SO₄), concentrated and the residue was purified by chromatography (silica gel , hexane/acetone 1:1) to give product

20 (158 mg, 83%).

Preparation of compound 21 A solution of 20 (130 mg, 0.2775 mmol) in pyridine (10 mL) was treated with acetic anhydride (524 mL, 5.549 mmol) at 25℃ for 48 h. The reaction was quenched with ice (10 g). The resulting solution was extracted with ethyl acetate $(3 \times 20 \text{ mL})$. The combined organic extracts were washed with 5% aqueous HCl, saturated aqueous NaH-CO₃, brine, dried (Na₂SO₄), concentrated and the residue was purified by chromatography (silica gel, hexane/acetone 4:1) to give 21 (80 mg, 52%) as a white solid, mp 211—213℃. ¹H NMR (500 MHz, CDCl₃): δ 6.03(d, J = 10.3 Hz, 1H, 10-H), 5.78(d, J =10.3 Hz, 1H, 9-H), 5.60(dd, J = 5.1, 2.1 Hz,1H, 2-H), 4.50(d, J = 12.0 Hz, 1H, 20-H), 4.06(d, J = 12.0 Hz, 1H, 20-H), 3.81(br. s, 1H, 5-H), 3.19(s, 1H, 4-OH), 3.13(d, J = 19.7 Hz, 1H, 14-H), 3.09(d, J = 5.0 Hz, 1H, 3-H), 2.73(dd, J = 19.8, 6.9 Hz, 1H, 14-H), 2.54(br. s,1H, 5-OH), 2.21(s, 3H, Ac), 2.19(s, 3H, Ac),2.14(dd, J = 6.7, 1.8 Hz, 1H, 1-H), 2.09(s, 3H,Ac), 2.08(s, 3H, Ac), 2.03(s, 3H, 18- CH_3), 1.87—1.81(m, 1H, 7-H), 1.79—1.76(m, 1H, 6-H), $1.76(s, 3H, 16-CH_3)$, 1.69-1.62(m, 1H, 7-Hor 6-H), 1.60—1.51(m, 1H, 6-H or 7-H), 1.14 (s, 3H, 17-CH₃), 0.85(s, 3H, 19-CH₃). FAB MS m/z: 591 (M + K⁺). Anal. $C_{28}H_{40}O_{11}$. Calcd.: C, 60.86; H, 7.30. Found: C, 60.58; H, 7.33.

Preparation of compound 22 To a stirred solution of **21** (264 mg, 0.478 mmol) in pyridine (10 mL) at 25°C was added methanesulfonyl chloride slowly and stirring was continued overnight. Then the reaction mixture was poured into ice-water (40 mL), extracted with ethyl acetate $(2 \times 40 \text{mL})$. The organic extracts were washed with dilute HCl, aqueous NaHCO3, brine, dride over Na₂SO₄ and concentrated to give crude product, which was purified by column chromatography (hexane/acetone 2:1) to yield 22 (249 mg, 83%), mp 230 °C (dec.). 1 H NMR (500 MHz, CDCl₃): δ 6.02(d, J = 10.4Hz, 1H, 10-H), 5.79(d, J = 10.4Hz)Hz, 1H, 9-H), 5.62(dd, J = 4.8, 1.9 Hz, 1H, 2-H), 4.97 (dd, J = 3.2, 2.1 Hz, 1H, 5-H), 4.73(d, J = 12.3 Hz, 1H, 20-H), 3.98(d, J = 12.3)Hz, 1H, 20-H), 3.19(s, 1H, 4-OH), 3.08(d, J =19.8 Hz, 1H, 14-H), 3.00(s, 3H, CH₃SO₂-), 2.97 (d, J = 4.7Hz, 1H, 3-H), 2.75(dd, J = 19.8, 6.8)Hz, 1H, 14-H), 2.25(s, 3H, Ac), 2.17(s, 3H,

Ac), 2.15(dd, J = 6.7, 1.8 Hz, 1H, 1-H), 2.12(s, 3H, Ac), 2.09(s, 3H, Ac), $2.04(s, 3H, 18-CH_3)$, 2.04-2.00(m, 1H, 6-H), 1.88-1.85(m, 1H, 6-H), $1.76(s, 3H, 16-CH_3)$, 1.79-1.70(m, 2H, 7-H), $1.16(s, 3H, 17-CH_3)$, $0.88(s, 1H, 19-CH_3)$. FAB MS m/z: $631(M+H^+)$. Anal. $C_{29}H_{42}-O_{13}S$. Calcd. C, 55.23; H, 6.71; S, 5.08. Found: C, 55.41; H, 6.81; S, 5.28.

Preparation of compound 23 To a solution of 19 (60 mg, 0.094 mmol) in pyridine (1.2 mL) was added methanesulfonyl chloride (33 mL, 0.421 mmol) and the reaction mixture was stirred at 25 °C for 24 h, then quenched with ice (5 g), extracted with ethyl acetate (3 × 10 mL). The combined organic extracts were washed with saturated aqueous CuSO₄, dried over Na₂SO₄, concentrated and purified by preparative TLC (silica gel, petroleum ether/acetone 2:1) to yield 23 (48 mg, 72%). $R_f = 0.5$.

Preparation of compound 24 Method A. To a solution of 22 (249 mg, 0.395 mmol) in a mixed solvent (methanol 30 mL, CH_2Cl_2 5 mL) at 0°C was added K_2CO_3 (109 mg, 0.790 mmol in 1.5 mL of H_2O) dropwise. The reaction mixture was stirred at 0°C for 30 min, then HCl (1 N, 1.58 mL) was added slowly, diluted with brine (50 mL), and extracted with $CH_2Cl_2(3 \times 50 \text{ mL})$. The combined extracts were washed with brine, dried over Na_2SO_4 , concentrated and chromatographed (hexane/acetone 1:1) to give product 24 (182 mg, 85%).

Method B. A solution of **23** (33 mg, 0.046 mmol) in methanol (1.2 mL) was treated with aqueous K_2CO_3 (1 N, 0.138 mL, 0.069 mmol) at -10 to -5°C for 40 min, then neutralized with 0.5% agucous HCl. After being concentrated, the residue was taken up with ethyl acetate (20 mL), washed with brine, dried over Na2SO4, concentrated and chromatographed (hexane/acetone 2:1) to give **24** (20 mg, 76 %). ¹H NMR (500 MHz, CDCl₃): δ 5.98(d, J = 10.4 Hz, 1H, 10-H), 5.72(d, J = 10.4 Hz, 1H, 9-H), 4.86(t, J = 2.7 Hz, 1H, 5-H), 4.46(s, 1H, OH), 4.25-4.30 (m, 2H, 2-H and 20-H), 4.18 (d, J = 9.1Hz, 1H, OH), 3.66(dd, J = 11.0, 5.5 Hz, 1H,20-H), 3.04(s, 4H, SO₂CH₃ and OH), 2.96(d, J =20.0 Hz, 1H, 14-H), 2.79(dd, J = 20.0, 6.7 Hz,1H, 14-H), 2.68(d, J = 4.9 Hz, 1H, 3-H), 2.30(dd, J = 6.7, 1.8 Hz, 1H, 1-H), 2.22(s, 3H, 18CH₃), 2.09(s, 3H, Ac), 2.03(s, 3H, Ac), 2.05—2.00(m, 1H, 6-H), 1.90—1.85(m, 1H, 6-H), 1.75—1.65(m, 2H, 7-H), 1.68(s, 3H, 16-CH₃), 1.18(s, 3H, 17-CH₃), 0.92(s, 3H, 19-CH₃). FAB MS m/z: 585 (M+K⁺).

Preparation of compound 25 A solution of 24 (8 mg, 0.015 mmol) in toluene (0.5 mL) was treated with DBU (3 mL), 0.022 mmol) at 105 ℃ for 1.5 h, then cooled to 25°C and diluted with cthyl acetate (10 mL). The resulting solution was washed with 0.5% aqueous HCl, saturated aqueous NaHCO₃, and brine, dried over Na₂SO₄, concentrated and purified by preparative TLC (silica gel, hexane/acetone 3:2) to give 25 (3 mg, 46%). ¹H NMR (500 MHz, CDCl₃): δ 5.92 (d, J = 10.4 Hz, 1H, 10-H), 5.82(d, J = 10.4)Hz, 1H, 9-H), 4.75(d, J = 8.6 Hz, 1H, 20-H), 4.72(dd, J = 9.2, 3.1 Hz, 1H, 5-H), 4.38(d, J)= 8.6 Hz, 1H, 20-H), 4.37---4.30(m, 1H, 2-H),3.68(s, 1H, OH), 2.81-2.72(m, 2H, 14-H),2.63(d, J = 9.1 Hz, 1H, OH), 2.29-2.22(m,1H, 1-H), 2.20-2.10(m, 1H, 6-H), 2.12(s, 1H, 6-H) Λc), 2.13(d, J = 5.5 Hz, 1H, 3-H), 2.05(s, 3H, $18-CH_3$), 2.04(s, 3H, Ac), 1.95-1.88(m, 1H, 6-1.88)H), $1.72(s, 3H, 19-CH_3)$, $1.35(s, 3H, 17-CH_3)$, 1.35-1.25(m, 2H, 7-H), 1.20(s, 3H, 16-CH₃).¹³CNMR(125MHz, CDCl₃): δ200.5, 170.5, 169.5, 150.8, 137.6, 87.4, 81.9, 76.6, 75.3, 73.0, 69.4 50.2, 48.7, 41.7, 38.0, 37.7, 35.0, 28.7, 26.8, 25.0, 20.9, 20.8, 16.6, 13.6. HR FAB MS Calcd. for $C_{24}H_{34}O_8(M + H^+)$: 451.2326. Found: 451.2325.

Compound 27 To a stirred solution of 26 (200 mg, 0.3263 mmol) and DMAP (797 mg, 6.526 mmol) in CH₂Cl₂ (8 mL) was added acetic anhydride (1.23 mL, 13.052 mmol). The reaction mixture was stirred at 35℃ for 5 h. The reaction was quenched with ice (2 g). The resulting solution was diluted with CH2Cl2 (40 mL), then the organic layer was washed with 5% aqueous HCl, brine, saturated aqueous NaHCO3, brine respectively. The organic layer was dried over Na2SO4, concentrated and the residue was purified by chromatography (silica gel, hexane/ Et₂O 2:1) to give 27 (127 mg, 56%) as a white solid, mp 68-71°C. ¹H NMR $(500 \text{MHz}, \text{CDCl}_3): \delta[8.12(d^*, J = 7.3 \text{Hz}) + 8.08]$ $(d^*, J = 7.3 \text{ Hz}), 2H, Ph-o], 7.59(t^*, J = 7.4)$ Hz, 1H, Ph-p), 7.48(t^* , J = 7.3 Hz, 2H, Ph-m),

6.00(dd, J = 12.2, 5.5Hz, 1H, 10-H), 5.77-5.70(m, 1H, 2-H), 5.02(d, J = 9.2 Hz, 1H, 5-H),[4.49(d, J = 8.0 Hz,) + 4.46(d, J = 7.9 Hz),1H, 20-H], 4.13-4.08(m, 2H, 20-H and 14-H), [12.14(s) + 5.23(s) + 3.68(d, J = 15.4 Hz) +3.55(d, J = 15.6 Hz), 2H, 4-COCH₂CO], [2.80](d, J = 6.0 Hz) + 2.75(d, J = 5.7 Hz), 1H, 3H], 2.56-2.48 (m, 1H, 9-H), 2.35-2.18 (m, 3H, 6-H and 2×13 -H), 2.08(s, 3H, Ac), 2.06(s, 3H, Ac)3H, Ac), 2.03-1.97 (m, 1H, 6-H), 1.95-1.89(m, 4H, 18-CH₃ and 6-H), [1.86(d, J = 3.1 Hz)]+ 1.85(d, J = 3.1 Hz), 1H, 1-H], 1.72(s, 3H,19-CH₃), 1.60—1.58(m, 2H, 9-H and 7-H), [1.40 (s) + 1.37(s), 3H, 17-CH₃], [1.12(s) + 1.10(s), 3H, 16-CH₃], 0.72(q, J = 8.0 Hz, 9H, TES-CH₃), 0.44—0.31(band, 6H, TES-CH₂). FAB MS m/z: 735(M + K⁺). Anal. $C_{39}H_{56}O_{9}Si$. Calcd.: C, 67.21; H, 8.10. Found: C, 67.49; H, 8.36. (*: with fine structure).

Preparation of compound 29 To a solution of 26 (500 mg, 0.8158 mmol) in THF (13.5 mL) was added lithium bis(trimethyl) silylamide (LiHMDS, 979 mL, 0.979 mmol, 1.0 M solution in THF) under N2 at 0 °C for 0 . 5 h , followed by acetyl chloride (7 0 μ L , 0.979 mmol). The solution was stirred at 0°C for 1.5 h. The reaction was quenched with saturated aqueous NH₄Cl (6 mL). The resulting solution was extracted with ethyl acetate $(4 \times 50 \text{ mL})$. The combined organic extracts were washed with saturated aqueous NaHCO₃, brine, dried (Na₂SO₄), concentrated and the residue was purified by chromatography (silica gel, hexane/ $Et_2O 2:1$ to 1:1) to give **29** (270 mg, 50.5%) as a white solid and recovered 26 (170 mg, 34%). 29: mp 68—70°C. ¹H NMR (500 MHz, CDCl₃): δ8.12(d*, J = 6.7 Hz, 2H, Ph-o), $7.58(t^*, J = 6.7 \text{ Hz}$, 1H, Ph-p), 7.47(t^* , J = 8.0 Hz, 2H, Ph-m), 6.00 (dd, J = 12.2, 4.9 Hz, 1H, 10-H), 5.72(dd, J =6.1, 3.1 Hz, 1H, 2-H), 4.96(d, J = 8.5 Hz, 1H,5-H), 4.47(d, J = 7.7 Hz, 1H, 20-H), 4.15(dd,J = 8.6, 3.7 Hz, 1H, 14-H), 4.09(d, J = 7.7 Hz, 1H, 20-H), 2.74(d, J = 5.5 Hz, 1H, 3-H), 2.51 (dd, J = 14.7, 12.8 Hz, 1H, 9-H), 2.39(dd, J =18.9, 3.7 Hz, 1H, 13-H), 2.34(dd, J = 18.9, 8.5 Hz, 1H, 13-H), 2.28(s, 3H, 4-Ac), 2.24—2.16 $(m, 2H, 2 \times 6-H), 2.06(s, 3H, 10-Ac), 2.04$ 1.96(m, 1H, 7-H), 1.93(s, 3H, 18-CH₃), 1.92-

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1.87(m, 1H, 7-H), 1.86(d, J = 3.1 Hz, 1H, 1-H), 1.72(s, 3H, 19-CH₃), 1.56(dd, J = 14.7, 4.9 Hz, 1H, 9-H), 1.36(s, 3H, 17-CH₃), 1.12(s, 3H, 16-CH₃), 0.72(t, J = 8.0Hz, 9H, TES-CH₃), 0.45 —0.31(m, 6H, TES-CH₂). FAB MS m/z: 693(M + K⁺). Anal. C_{37} H₅₄O₈Si. Calcd: C, 67.86; H, 8.31. Found: C, 67.76; H, 8.43. (*: with fine structure).

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